PAPER • OPEN ACCESS

Analysis of the radiation parameters of pumice constituent compounds using XCOM

To cite this article: R Wirawan et al 2021 J. Phys.: Conf. Ser. 1816 012116

View the article online for updates and enhancements.

You may also like

- X-ray fluorescent (XRF) configuration for the measurement of mass attenuation coefficients at low energy photons S N A Abdullah, M F Mohd Yusof, N A Kabir et al.
- <u>The comparative study of the radiation</u> shielding of PbO-Li₂O-B₂O₃ glass system by using FLUKA to XCOM and <u>experimental data</u> C Sriwunkum, T Nutaro and A Saiz
- The comparative studies of gamma-ray shielding properties of the PbO-BaO-B₂O₂ glass system by using FLUKA code to XCOM program and accessible experimental data C Mutuwong, T Nutaro and A Saiz



IOP ebooks[™]

Bringing together innovative digital publishing with leading authors from the global scientific community.

Start exploring the collection-download the first chapter of every title for free.

Analysis of the radiation parameters of pumice constituent compounds using XCOM

R Wirawan¹, N Qomariyah¹, T Ardianto¹, and D W Kurniawidi¹

¹Department of Physics, Faculty of Mathematics and Natural Science, University of Mataram, Jl. Majapahit 62 Mataram 83125, Indonesia

E-mail: rwirawan@unram.ac.id

Abstract. In this study, analysis of radiation parameters (linear attenuation coefficient, mean free path and half value layer) of pumice constituent compounds was carried out. Analyzes were done on the compounds of Al₂O₃, Fe₂O₃, SiO₂, CaO, K₂O and Na₂O with a theoretical approach using the XCOM program for gamma photon energies 57 Co (0.122 MeV; 0.1365 MeV), 133 Ba (0.356 MeV), 54 Mn (0.835 MeV), 137 Cs (0.662 MeV), 60 Co (1.173 MeV; 1.333 MeV), 65 Zn (1.1155 MeV), 22 Na (0.511 MeV; 1.275 MeV) and 40 K (1.461 MeV). The results show that the Fe₂O₃ compound has the largest linear attenuation coefficient, while the largest for MFP and HVL is Na₂O among other compounds.

1. Introduction

The use of radiation is currently growing rapidly in the many fields of medicine, agriculture, security and industry. The important thing that must be considered in the utilization of radiation is avoiding the risk of excessive radiation exposure. Therefore, a type of protective material is used to reduce the radiation intensity. The use of certain materials as radiation shielding becomes an important issue in the development of radiation protection systems, especially to replace lead shield which is toxic [1, 2]. There are several investigations of various materials for radiation shielding such as basalt, marble, granite and limestone [3], gadolinium oxide and oxyfluoride glass [4], bismuth borosilicate glass [5], and pumice [6].

Pumice is a rock that results from volcanic eruptions and its utilization based on the characteristics of its properties, both physical and chemical properties. Pumice can be used as a brick material [7], fine aggregate as a base for making lightweight concrete and as a substitute for cement [8, 9]. The use of pumice can affect the strength and durability characteristics of concrete, especially in high strength concrete [10]. Despite the lead, concrete is one of the composites which applied as shielding wall [11, 12]. To develop the potential of pumice as a radiation shielding material, it is necessary to know the radiation parameters characteristic of its constituent compounds. A good radiation shielding has small half-value layer (HVL) and the mean free path (MFP) parameters. Those values depend on the linear attenuation coefficient and which determined based on the mass absorption coefficient value of the shielding material compounds.

In this study, the analysis of radiation parameters of the pumice constituent compounds was carried out, namely the linear attenuation coefficient, the mean free path and the half-value layer. The analysis used the mass absorption coefficient value which is determined from the photon cross-section (XCOM) program.

Content from this work may be used under the terms of the Creative Commons Attribution 3.0 licence. Any further distribution of this work must maintain attribution to the author(s) and the title of the work, journal citation and DOI. Published under licence by IOP Publishing Ltd 1

The 10th International Conference on Th	heoretical and Applied Physics (IC	CTAP2020)	IOP Publishing
Journal of Physics: Conference Series	1816 (2021) 012116	doi:10.1088/17	42-6596/1816/1/012116

2. Method

In this study, the pumice constituent compounds analyzed were six compounds with the largest concentration, namely Al_2O_3 , Fe_2O_3 , SiO_2 , CaO, K_2O , and Na_2O [8, 9,13]. The atomic numbers of the constituent elements are shown in Table 1.

 Table 1. The atomic number and mass of the elements of the pumice constituent compounds [14]

Element	Atomic number (Z)	Atomic mass (A)
0	8	15.9994
Al	13	26.9815
Si	14	28.0855
Κ	19	39.0983
Ca	20	40.0780
Fe	26	55.8450
Na	11	22.9898

The mean atomic number (Z_{comp}) of a compound or mixture that makes up the pumice can be determined by the formula in equation (1) [15].

$$Z_{comp} = \frac{\sum_{i=1}^{N} w_i Z_i^2 / A_i}{\sum_{i=1}^{N} w_i Z_i / A_i}$$
(1)

where *N* is the number of elements in the compound or mixture, $w_i = \frac{N_i A_i}{M}$ is the weight fraction of *i*th element, N_i is number of atoms, Z_i and A_i are the atomic number and the atomic mass of *i*th element. Table 2 shows the mean atomic number (Z_{comp}) of pumice constituent compounds which calculated using the equation (1).

Table 2. Mean atomic number (Z_{comp}) and density of pumice constituent compounds

Compounds	Atoms of co (Z : mass f	ompounds raction)	Z_{comp}	Density [16] (gr/cm ³)
Fe ₂ O ₃	8:0.30057	26:0.69943	20.315873	5.250
Al_2O_3	8:0.47075	13:0.52925	10.599996	3.965
CaO	8:0.28531	20:0.71469	16.571419	3.340
K ₂ O	8:0.16985	19:0.83015	17.086981	2.320
SiO_2	8:0.53257	14:0.46743	10.799970	2.320
Na ₂ O	8:0.25814	11:0.74186	8.280026	2.270

The linear attenuation coefficient (μ) is a radiation parameter that states the total interaction probability (photoelectric, Compton, pair production) of photons with the material, and its value varies depending on the material density (ρ). These coefficients can be determined based on the mass absorption coefficient in equation (2) [17].

$$\mu(E) = \left(\frac{\mu}{\rho}(E)\right)\rho\tag{2}$$

Mass absorption coefficients $\frac{\mu}{\rho}(E)$ of pumice constituent compound are calculated using the photon cross-section program (XCOM Ver. 3.1) with input parameters are element type and the number of

The 10th International Conference on T	Theoretical and Applied Physics (IC	CTAP2020)	IOP Publishing
Journal of Physics: Conference Series	1816 (2021) 012116	doi:10.1088/1742-	6596/1816/1/012116

atoms. The photon energies analyzed are the gamma photon energies from the following sources ⁵⁷Co (0.122 MeV; 0.1365 MeV), ¹³³Ba (0.356 MeV), ⁵⁴Mn (0.835 MeV), ¹³⁷Cs (0.662 MeV), ⁶⁰Co (1.173 MeV; 1.333 MeV), ⁶⁵Zn (1.1155 MeV), ²²Na (0.511 MeV; 1.275 MeV), and ⁴⁰K (1.461 MeV). Meanwhile, the mean free path (MFP) and half-value layer (HVL) parameters are determined using the following formula [12].

$$MFP = \frac{1}{\mu} \tag{3}$$

and

$$HVL = \frac{\ln 2}{\mu} \tag{4}$$

3. Result and Discussion

Figure 1 shows a graph of the total mass absorption coefficient of the XCOM output results for six pumice constituent compounds and the mass absorption coefficient of lead (PbO) which is generally used as a radiation shielding material in the photon energy range 0.01 - 20 MeV. The graph of the total mass absorption coefficient shown is a combination of photoelectric interactions, Compton (incoherent) scattering effect and pair production.



Figure 1. Graph of a mass absorption coefficient of pumice constituent compounds

If we observe the graphs in Figure 2(a) and Figure 2(c), it appears that the Fe_2O_3 compound has a higher coefficient value for photoelectric interaction and pair production. The high coefficient value can be explained based on the mean atomic number in Table 2 where the Fe_2O_3 compound has an atomic number greater than the five other forming compounds, is about 20.315873. Meanwhile, the graph of the mass absorption coefficient for the Compton scattering interaction (Figure 2b) shows that the SiO2 compound is higher than the other five compounds. Table 3 shows the mass absorption coefficient for gamma energies from radioactive sources ${}^{57}Co$, ${}^{133}Ba$, ${}^{54}Mn$, ${}^{137}Cs$, ${}^{60}Co$, ${}^{65}Zn$, ${}^{22}Na$ and ${}^{40}K$.



Figure 2. Graph of the mass absorption coefficient of (a) photoelectric effect, (b) Compton scattering, and (c) pair production

Gamma photon	Mass	Mass coefficient absorption (μ/ρ) of pumice compound (cm^2/g)					
energies (MeV)	Fe_2O_3	Al_2O_3	CaO	K_2O	SiO ₂	Na ₂ O	
0.122 (⁵⁷ Co)	0.2076	0.1422	0.1717	0.1664	0.1457	0.1389	
0.1365 (⁵⁷ Co)	0.1817	0.1368	0.1580	0.1534	0.1399	0.1340	
0.356 (¹³³ Ba)	0.0975	0.0978	0.1005	0.0982	0.0996	0.0965	
0.511 (²² Na)	0.0830	0.0846	0.0864	0.0845	0.0862	0.0835	
$0.662 (^{137}Cs)$	0.0739	0.0756	0.0772	0.0755	0.0770	0.0746	
0.835 (⁵⁴ Mn)	0.0663	0.0680	0.0693	0.0678	0.0693	0.0672	
1.115 (⁶⁵ Zn)	0.0575	0.0591	0.0603	0.0589	0.0602	0.0584	
1.173 (⁶⁰ Co)	0.0561	0.0577	0.0588	0.0575	0.0587	0.0569	
1.275 (²² Na)	0.0538	0.0553	0.0563	0.0551	0.0563	0.0546	
1.333 (⁶⁰ Co)	0.0526	0.0540	0.0551	0.0539	0.0550	0.0533	
1.461 (⁴⁰ K)	0.0503	0.0516	0.0526	0.0515	0.0525	0.0509	

Table 3. Mass coefficient absorption of pumice constituent compounds for specific gamma energy

Figure 3 shows a graph of the linear attenuation coefficient for the pumice constituent compounds which calculate using equation (2). In the graph, it appears that lowering the linear attenuation coefficient due to the increasing energy of gamma photons. Since the linear attenuation coefficient depends on the material density parameter, and Fe_2O_3 has a greater density compared to the density of the five other constituent compounds. So, Fe_2O_3 has a higher linear attenuation coefficient value than the other five compounds as shown in the graph.

The MFP value indicates a distance between two successive interactions. The MFP will get longer along with the increasing of incoming photon energy. Na_2O has a shorter MFP than other compounds. In addition, an increase of density value of a compound, MFP of the compound is shorter as shown in Figure 4. This is related to the number of atoms of material, where the high density material has a lot of atoms. These results are similar to the results of experimental investigations for MFP of concrete samples obtained by Agar et al. [12].

The HVL value states the thickness of a material needed to be able to pass 50 percent of the radiation intensity that penetrates the material. Table 4 shows the results of the calculation of the HVL value of the pumice constituent compounds for different photon energies. It is observed that the greater of incoming photon energy, requiring the thick material shield. The contribution of the radiation parameters of a compound to the total parameters of pumice radiation depends significantly on that compound content in the pumice. Pumice with a large enough Fe_2O_3 content will makes a significant contribution for attenuating photon radiation intensity.



Figure 3. Linear attenuation coefficient of pumice constituent compounds



Figure 4. Photon mean free path of pumice constituent compounds.

Table 4. Half-value la	ver (HVL) of	pumice constituent of	compounds for sp	pecific gamma energy

Gamma photon	HVL pumice constituent compounds (cm)					
energies (MeV)	Fe ₂ O ₃	Al_2O_3	CaO	K ₂ O	SiO ₂	Na ₂ O
0.122 (⁵⁷ Co)	0.6360	1.2294	1.2087	1.7955	2.0506	2.1984
0.1365 (⁵⁷ Co)	0.7266	1.2779	1.3135	1.9477	2.1356	2.2787
0.356 (¹³³ Ba)	1.3548	1.7871	2.0650	3.0431	2.9988	3.1639
0.511 (²² Na)	1.5907	2.0664	2.4011	3.5362	3.4680	3.6569
$0.662 (^{137}Cs)$	1.7875	2.3112	2.6896	3.9598	3.8791	4.0910
0.835 (⁵⁴ Mn)	1.9914	2.5697	2.9929	4.4060	4.3125	4.5473
1.115 (⁶⁵ Zn)	2.2949	2.9565	3.4428	5.0699	4.9613	5.2331
1.173 (⁶⁰ Co)	2.3534	3.0318	3.5318	5.1996	5.0881	5.3665
1.275 (²² Na)	2.4541	3.1624	3.6835	5.4223	5.3077	5.5976

The 10th International Conference on Theoretical and Applied Physics (ICTAP2020)IOP PublishingJournal of Physics: Conference Series**1816** (2021) 012116doi:10.1088/1742-6596/1816/1/012116

Gamma photon		HVL p	umice consti	tuent compou	unds (cm)	
energies (MeV)	Fe ₂ O ₃	Al_2O_3	CaO	K ₂ O	SiO ₂	Na ₂ O
1.333 (⁶⁰ Co)	2.5096	3.2349	3.7671	5.5451	5.4302	5.7257
1.461 (⁴⁰ K)	2.6274	3.3912	3.9462	5.8070	5.6909	6.0014

4. Conclusion

In this work, it was found that Fe_2O_3 has the largest linear attenuation coefficient compared to the other five of the pumice constituent compound. In addition, Na_2O have the largest HVL and MFP parameters. Contribution of those radiation compound parameters to the radiation shielding parameters of pumice depends on its composition in the pumice.

Acknowledgments

Authors are grateful to LPPM University of Mataram for the BLU (PNBP) 2020 funding research.

References

- [1] Wani A L, Ara A and Usmani J A 2015 Interdiscip Toxicology 8 55
- [2] AbuAlRoos N J, Amin N A B and Zainon R 2019 *Radiation Physics and Chemistry* **165** 108439
- [3] Obaid S S, Sayyed M I, Gaikwad D K and Pawar P P 2018 Radiation Physics and Chemistry 148 86
- [4] Shamshad L, Rooh G, Limkitjaroenporn P, Srisittipokakun N, Chaiphaksa W, Kim H J and Kaewkhao J 2017 Progress in Nuclear Energy 97 53
- [5] Cheewasukhanont W, Limkitjaroenporn P, Kothan S, Kedkaew C and Kaewkhao J 2020 *Radiation Physics and Chemistry* **172** 108791
- [6] Tapan M, Yalçın Z, Orhan İçelli O, Kara H, Orak S Özvan A and Depci T 2014 Annals of Nuclear Energy 65 290
- [7] Turhan S, Demir K and Karata M 2018 *Applied Radiation and Isotopes* **141** 95
- [8] Yaltay N, Ekinci C E, Cakir T and Oto B 2015 Progress in Nuclear Energy 78 25
- [9] Hossain K M A 2004 Cement and Concrete Research 34 283
- [10] Zeyad A M, Khan A H and Tayeh B A 2020 Journal of Materials Research and Technology 9 806
- [11] Gur A, Artig B, and Cakir T 2017 Physicochemical Problems of Mineral Processing 53 184
- [12] Agar O, Tekin H O, Sayyed M I, Korkmaz M E, Culfa O, and Ertugay C 2019 Results in Physics 12 237
- [13] Ismail A I M, El-Shafey O I, Amr M H A and El-Maghraby M S 2014 International Scholarly Research Notices 2014 259379
- [14] Lide D R 2004 Handbook of Chemistry and Physics-84th ed. (Boca Raton: CRC Press LLC)
- [15] Tsoulfanidis 1995 *Measurement and Detection of Radiation-2nd ed.* (United States of America: Taylor & Francis)
- [16] http://www.aqua-calc.com
- [17] Knoll, G.F. 2010 Radiation detection and measurement-4th ed. (United States of America: John Wiley & Sons, Inc.)